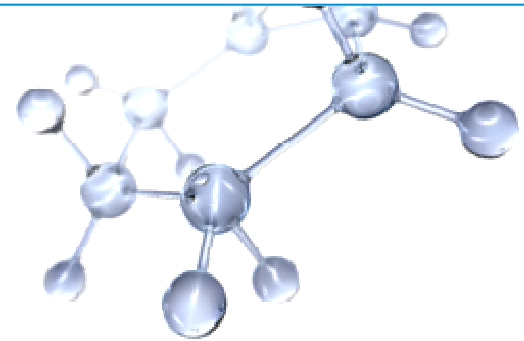


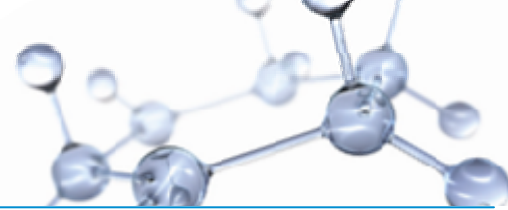


Assessing oil toxicity: methods & models



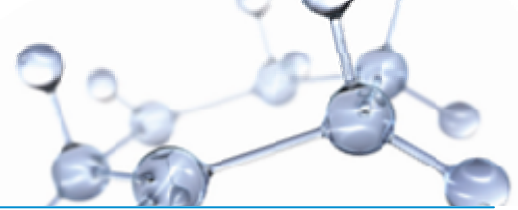
University of Florida Institute of Oceanography

Thomas F. Parkerton
September 27, 2011



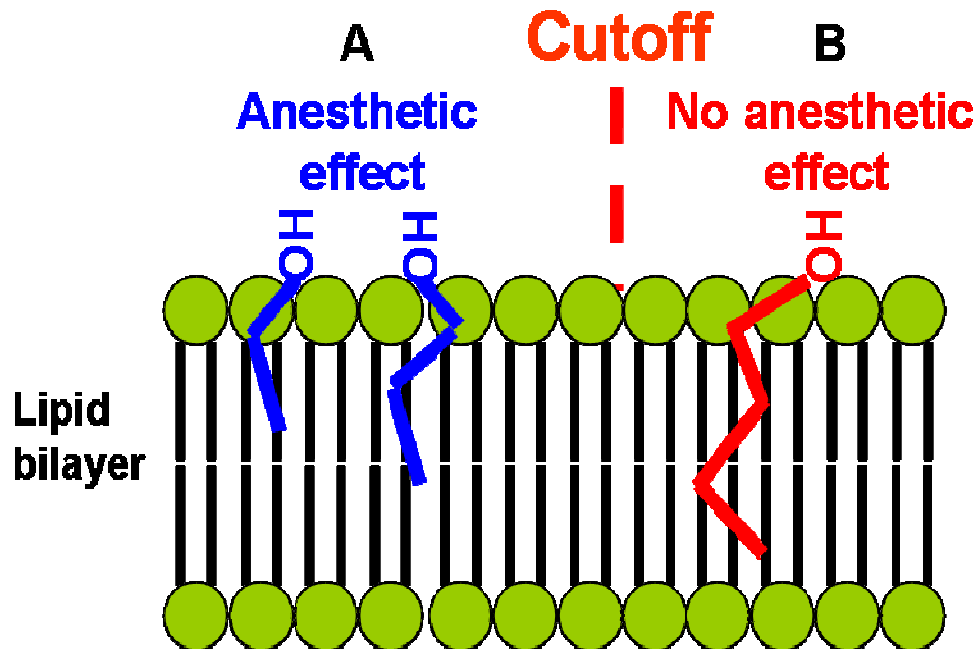
Outline

- Toxicity assessment of single hydrocarbons to aquatic/marine life
 - Target lipid model
- Methods for testing complex hydrocarbons, e.g. crude oil
 - Water Accomodated Fraction (WAF) test procedure
- Tools for predicting toxicity
 - Additive toxic unit model
 - Biomimetic extraction analysis
- Influence of chemical dispersants on oil toxicity
- Additional issues
 - Photo-enhanced toxicity
 - Bioaccumulation of PAHs in foodchain
- Summary & research needs

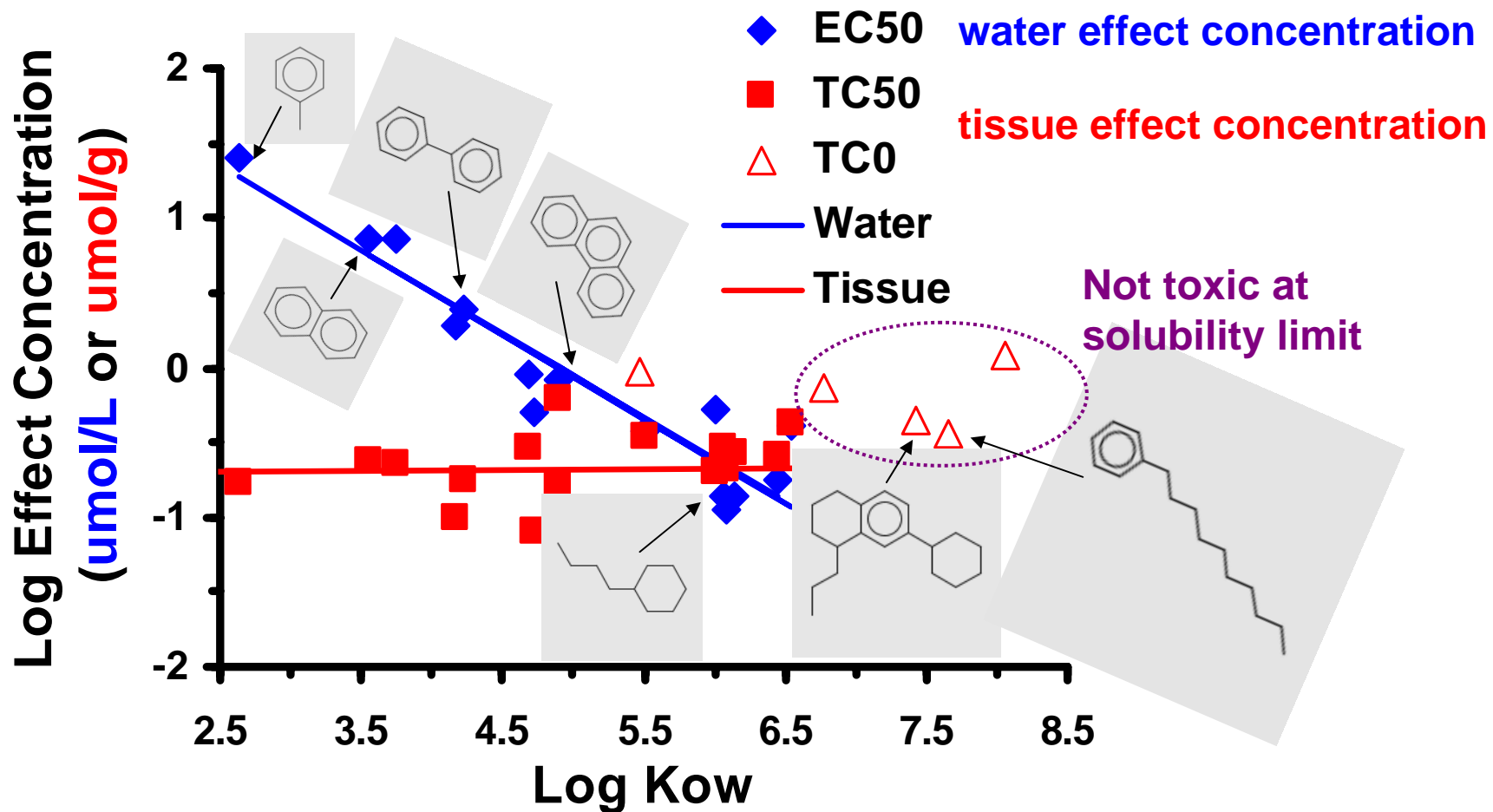
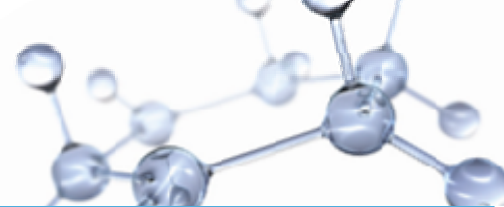


Narcosis

- Non-specific, perturbation of membrane function that results in decreased activity (e.g. ventilation, oxygen consumption, heart rate), immobilization and ultimately death to organisms
- Applicable to many classes of chemicals including hydrocarbons
 - Minimum level or “baseline” toxicity independent of exposure route
 - Shown to correlate with substance hydrophobicity until toxicity “cut-off”



Inhibition of Mussel Filtration Rate



Source:

Donkin et al. (1991) Sci. Tot. Env. 109:461; Smith et al. (2001) ET&C 20:2482

Predicting Narcosis using the Target Lipid Model



CTLBB for a chemical is determined as:

$$\text{CTLBB} = \text{LC}_{50} \times \text{KT}_{\text{L-W}} \quad (1)$$

Rearranging and taking logs:

$$\log(\text{EC}_{50}) = \log(\text{CTLBB}) - \log(\text{KT}_{\text{L-W}}) \quad (2)$$

Based on linear-free energy relationships:

$$\log(\text{KT}_{\text{L-W}}) = a_0 + a_1 \log(\text{K}_{\text{ow}}) \quad (3)$$

Substituting (3) into (2) yields:

$$\underbrace{\log(\text{EC}_{50})}_{\text{Response}} = \underbrace{\log(\text{CTLBB})}_{\text{Organism}} - \underbrace{a_0 - a_1 \log(\text{K}_{\text{ow}})}_{\text{Hydrocarbon}} \quad (4)$$

CTLBB = critical target lipid body burden (mmol/kg octanol)

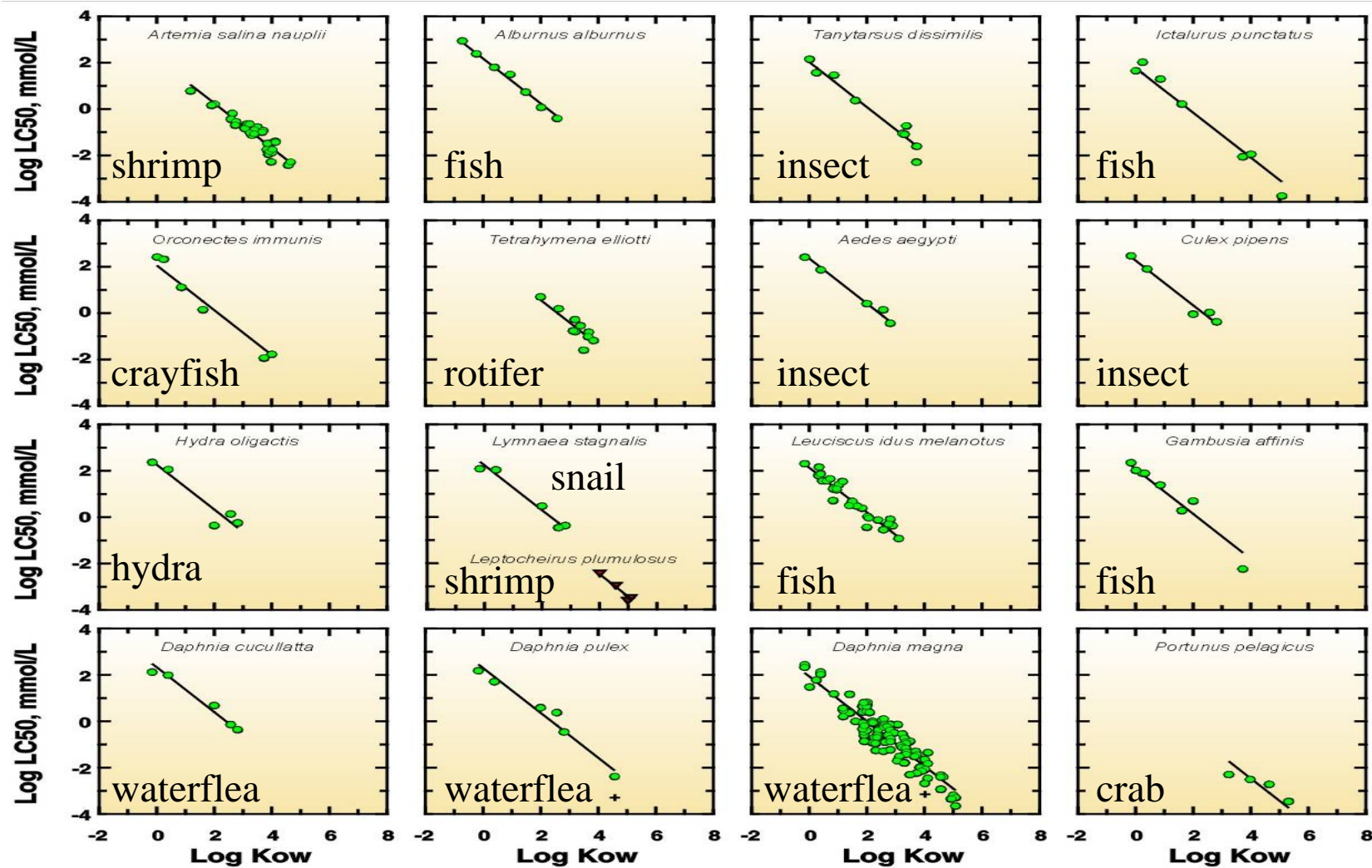
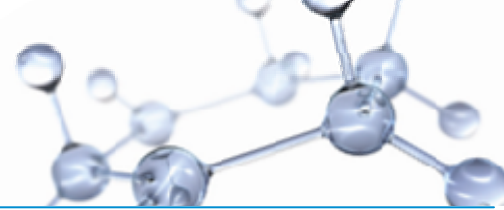
EC₅₀ = aqueous concentration that causes a 50% response (mmol/L)

KT_{L-W} = target lipid water partition coefficient (L/kg lipid)

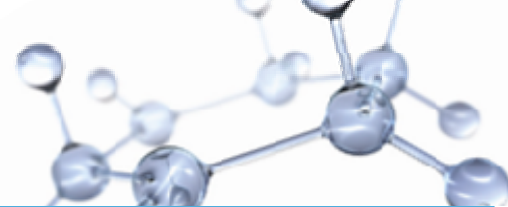
K_{ow} = octanol water partition coefficient (L/kg octanol)

a₀, a₁ = empirical constants that relate partitioning at target site to octanol

Calibration of TLM using Acute Toxicity Data Sets



Source: DiToro, McGrath, & Hansen (2000) ET&C 19:1951–1970.

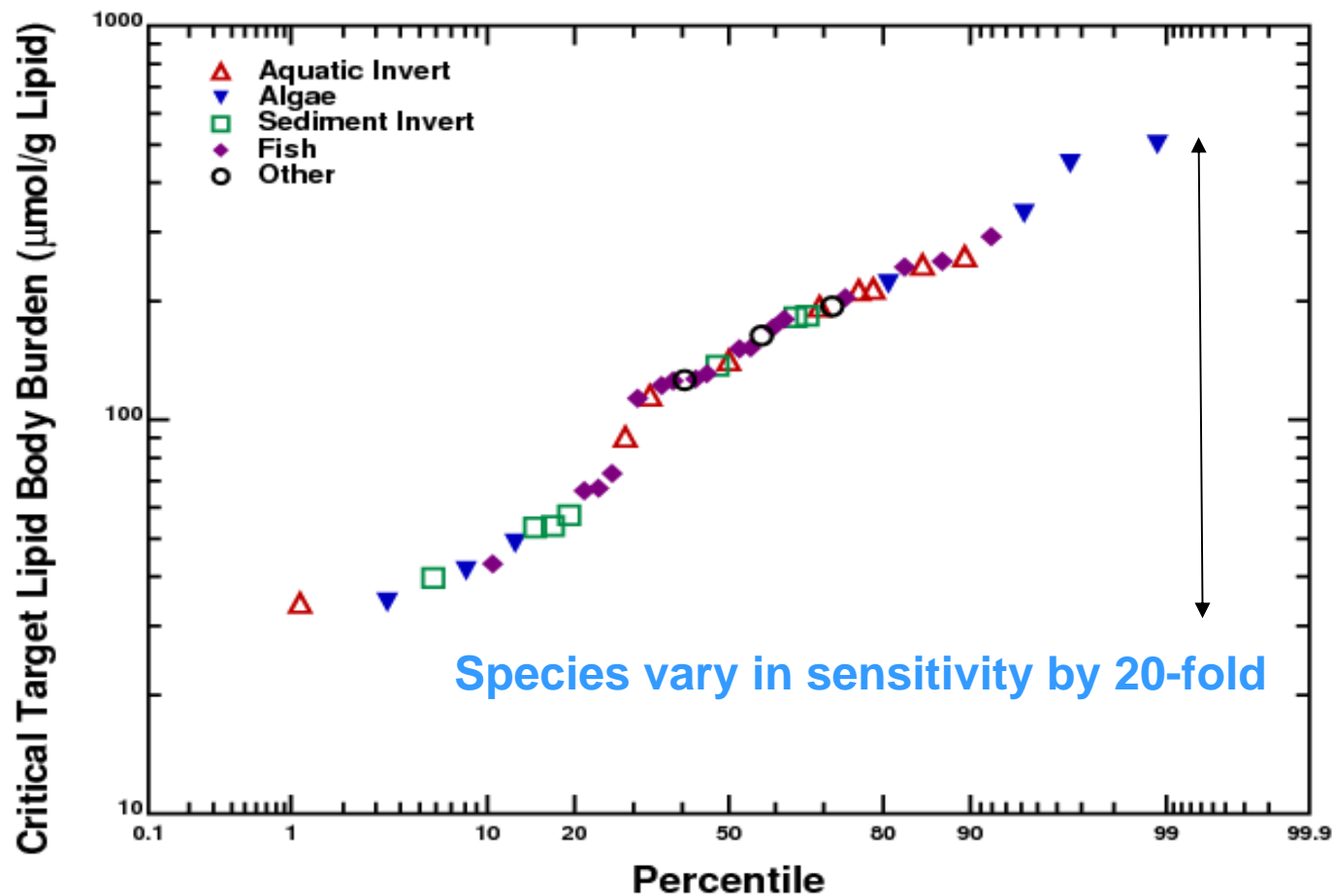


Results of TLM calibration

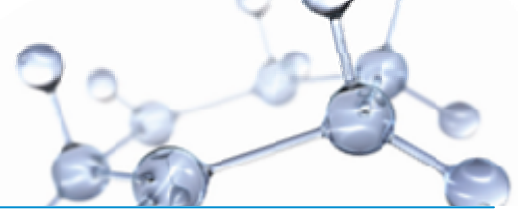
- Quantitative relationships developed for 56 species
 - amphibians, fish, invertebrates, algae, microbes / aquatic & marine
 - ca. 1000 reliable acute toxicity tests for 250+ chemicals
 - + aliphatic hydrocarbons, alcohols, ethers, ketones, mono-, and poly-aromatic hydrocarbons including halogenated structures than span a $\log(K_{OW})$ range from 0 to 6
- a_0 chemical class dependent
 - $a_0 = 0$ for most HCs (baseline); = 0.35 for PAHs (2X potency)
 - attributed to polar interactions that increase affinity for target site
- a_1 constant across narcotic chemicals!
 - $a_1 = 0.936$
- Intercept [$\log(CTLBB)$] is species-dependent
 - used to define species-sensitivity distribution (next slide)

Source: McGrath & DiToro (2009) ET&C28:1130

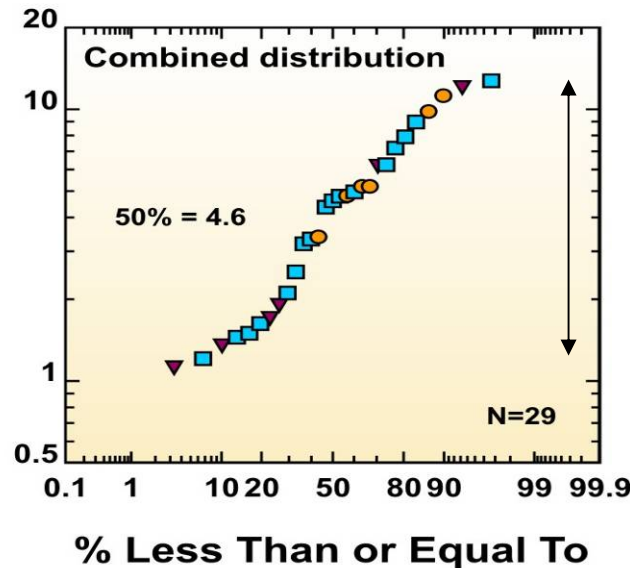
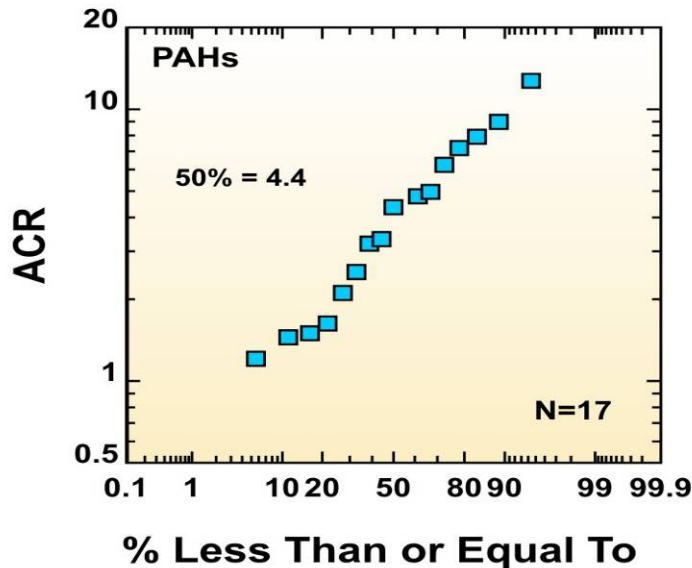
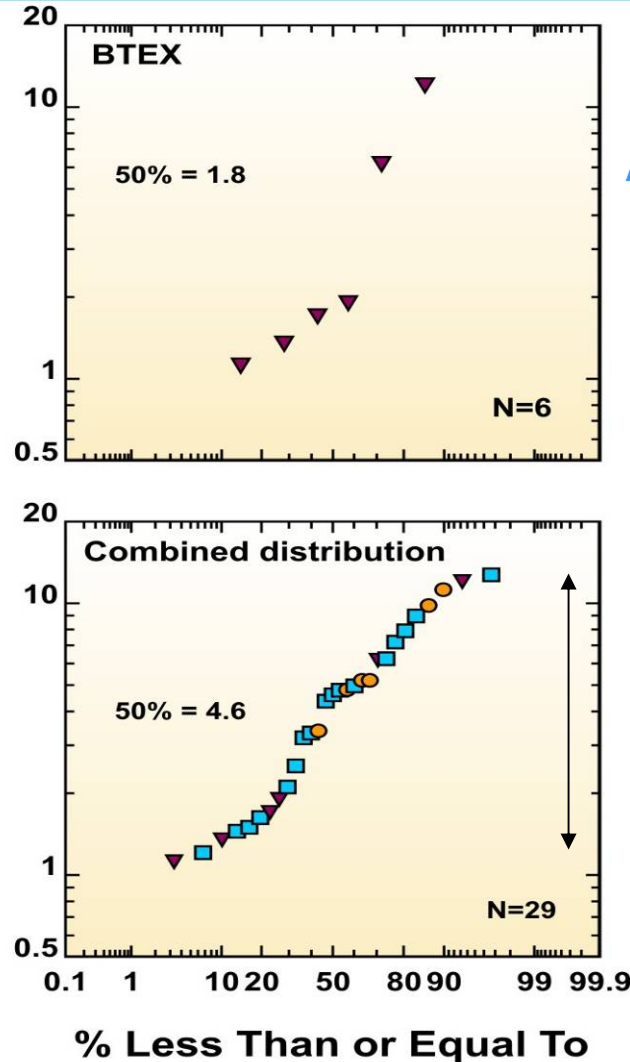
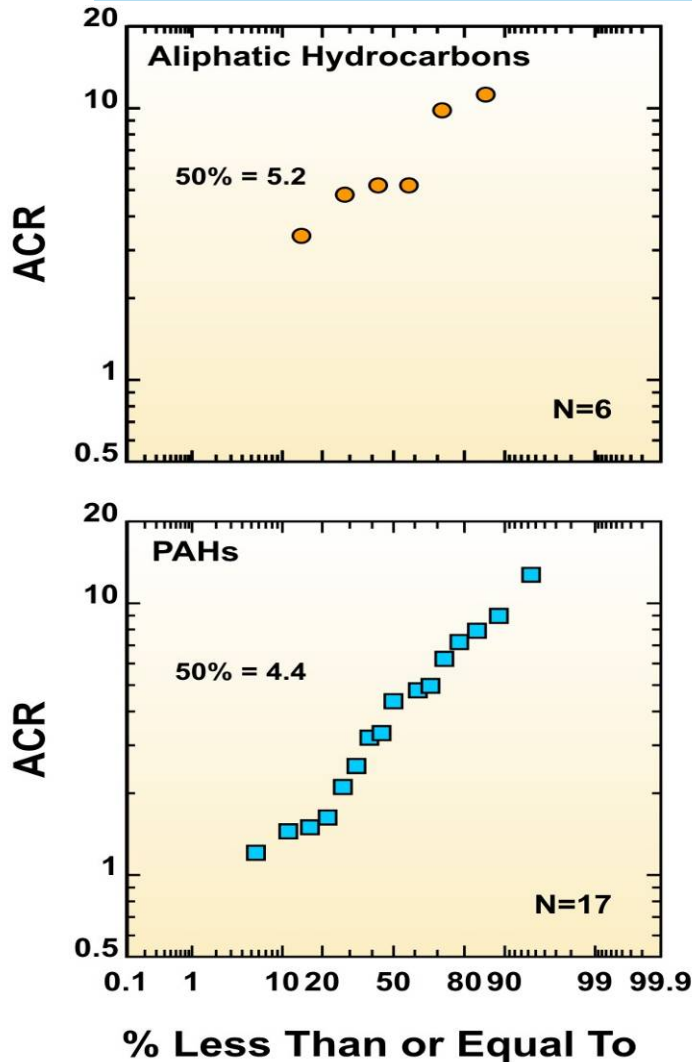
CTLBB Species - Sensitivity Distribution



Source: McGrath & DiToro (2009) ET&C 28:1130



Extrapolation to Chronic Effects

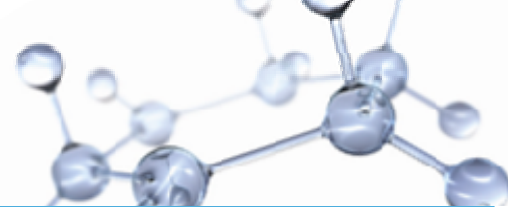


Acute to Chronic Ratio

$$ACR = \frac{\text{acute } L/EC_{50}}{\text{chronic } NOEC/EC_{10}}$$

For hydrocarbons
ACRs vary 10-fold

Source:
McGrath & DiToro
(2009) ET&C 28:1130



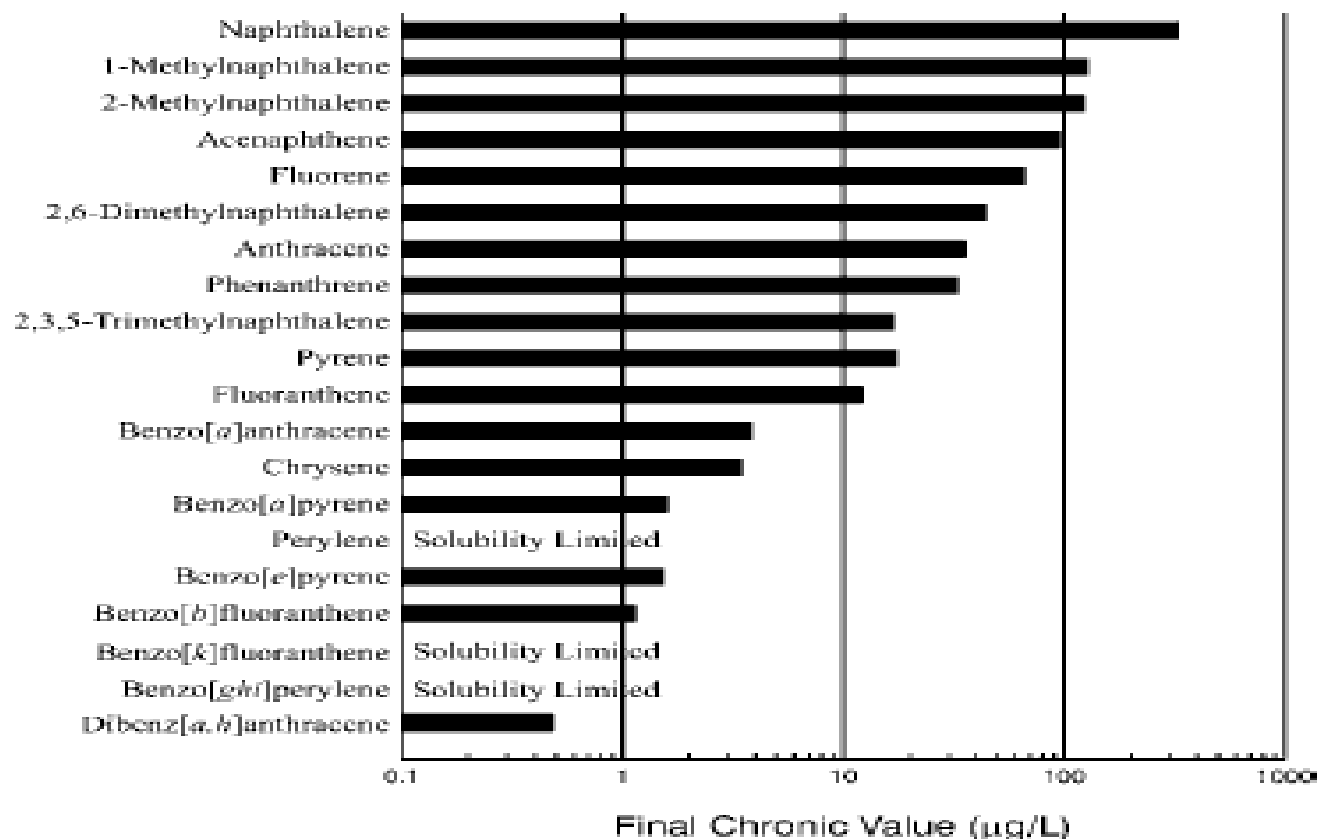
Derivation of Water Quality Criteria

- Final Chronic Value (mmol/L) is given by:

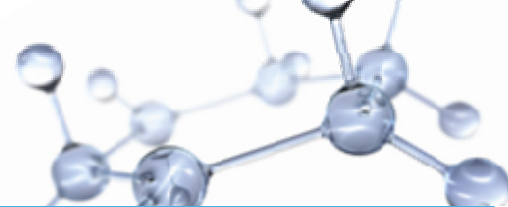
$$\log(\text{FCV}) = \log(\text{CTLBB}_{5\text{th}}) - 0.936 \log(K_{ow}) - a_0 - \log(\text{GMACR})$$

CTLBB5th = 5th percentile of CTLBB species-sensitivity distribution

GMACR = Geometric Mean Acute to Chronic Ratio



Source:
DiToro et al. (2007)
ET&C 26:24



Testing Complex Substances

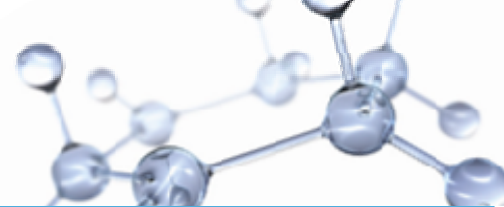
- Water Accommodated Fraction (WAF): An aqueous medium containing the fraction of the petroleum product that remains in the aqueous phase once mixing is terminated and phase separation has occurred
 - WAF = soluble phase (dissolved fraction) + droplets (colloidal fraction)
 - WAFs are prepared at multiple oil-water ratios (i.e. Loadings)
 - Test method described by OECD guidance document
 - + http://www.epa.gov/endo/pubs/ref-2_oecd_gd23_difficult_substances.pdf
- Practical Considerations:
 - How to add the test substance to dilution water?
 - How to mix?
 - How long to equilibrate?
 - How long for phase separation after mixing?
 - How to sample WAFs for testing?
 - How to expose test organisms and express test results?



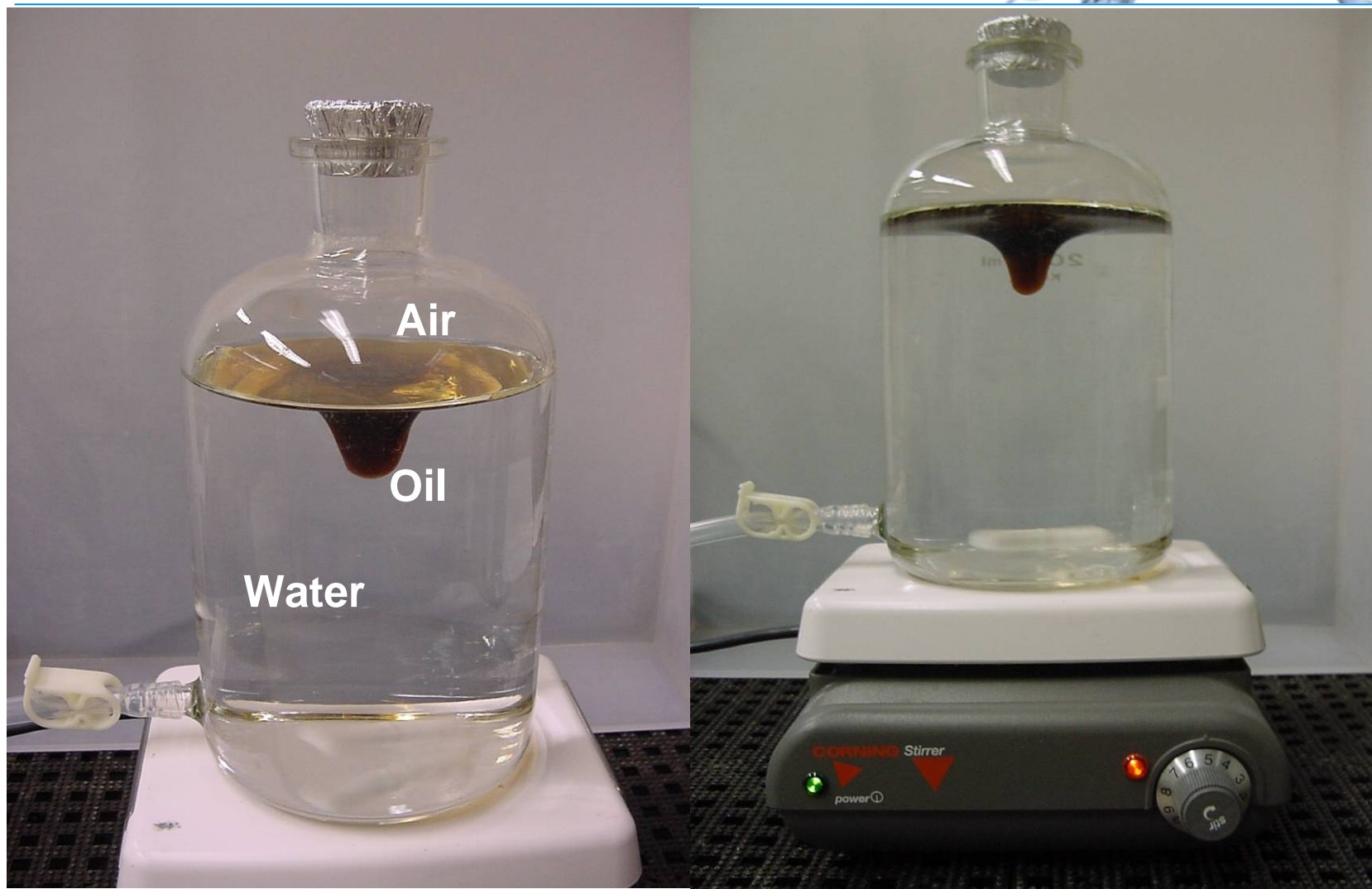
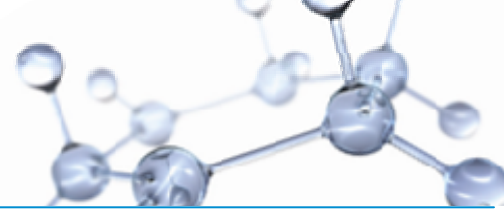
Outline of WAF Test Procedure

- Add a measured volume (liquids) or weight (solids) of substance to known volume of water in a sealed test vessel
 - Contains 5-10% headspace to allow mixing & includes Teflon coated stir-bar
 - Equipped with port at bottom for sampling WAFs with low density (floating) or glass siphon tube in middle for sampling high density (sinking) products
- Stir oil-water solution on magnetic stir plate at a rate that provides good mixing but prevents emulsion formation
 - Use mixing rate that creates $< 10\%$ vortex of static depth of oil-water solution
 - Typically stir at room temperature (22 ± 2 °C)
- Continue mixing until equilibrium is obtained
 - Take periodic samples for chemical analysis
 - + TOC, Solvent extraction coupled with UV Spectroscopy/GC-FID or MS
 - + Solid phase microextraction (SPME) coupled with GC-FID or MS
 - 48-96 hrs generally sufficient for most complex petroleum substances

WAF Preparation of Liquids



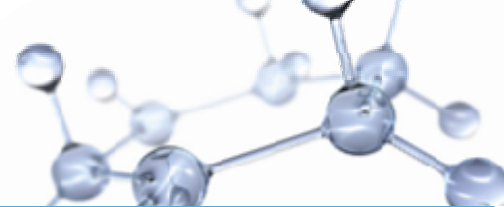
WAF Vessel / Mixing



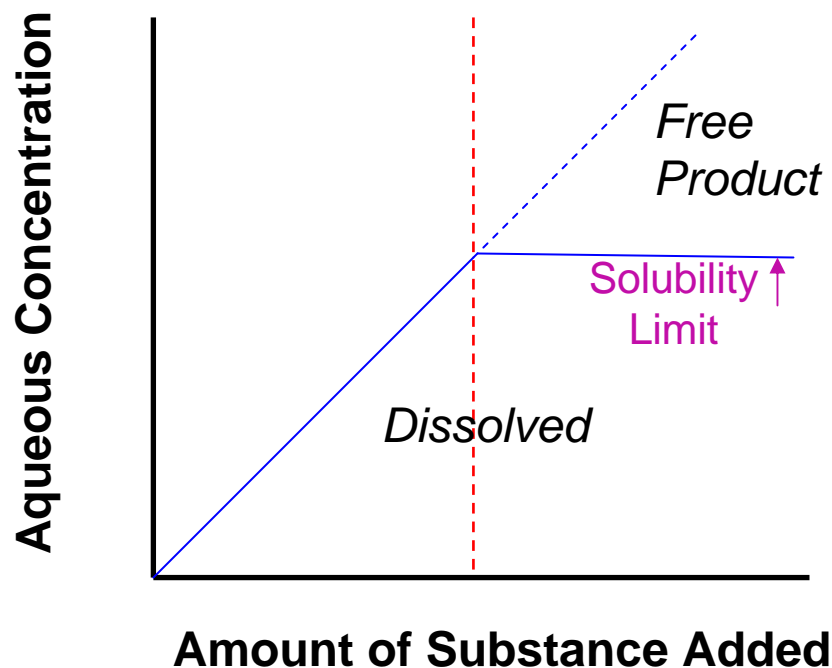
Preparation of WAFs for Toxicity Testing



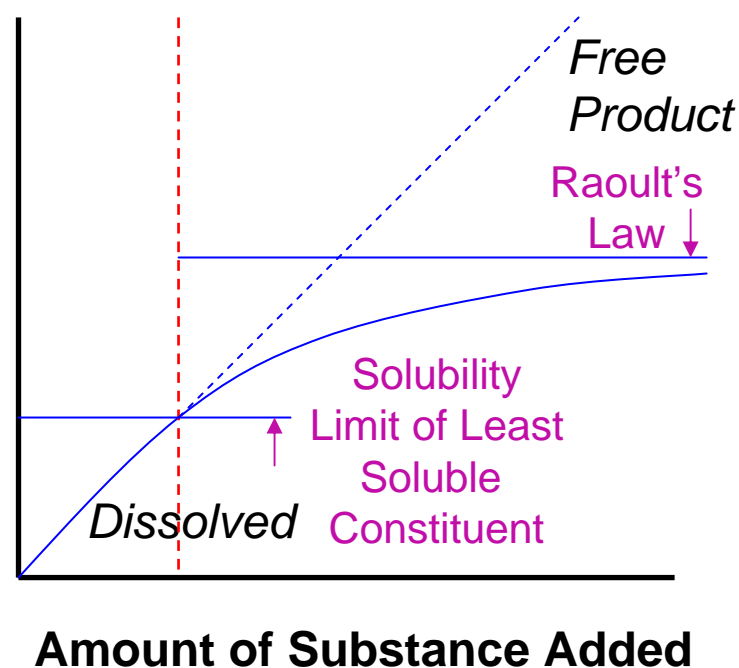
Aqueous Solubility Behavior



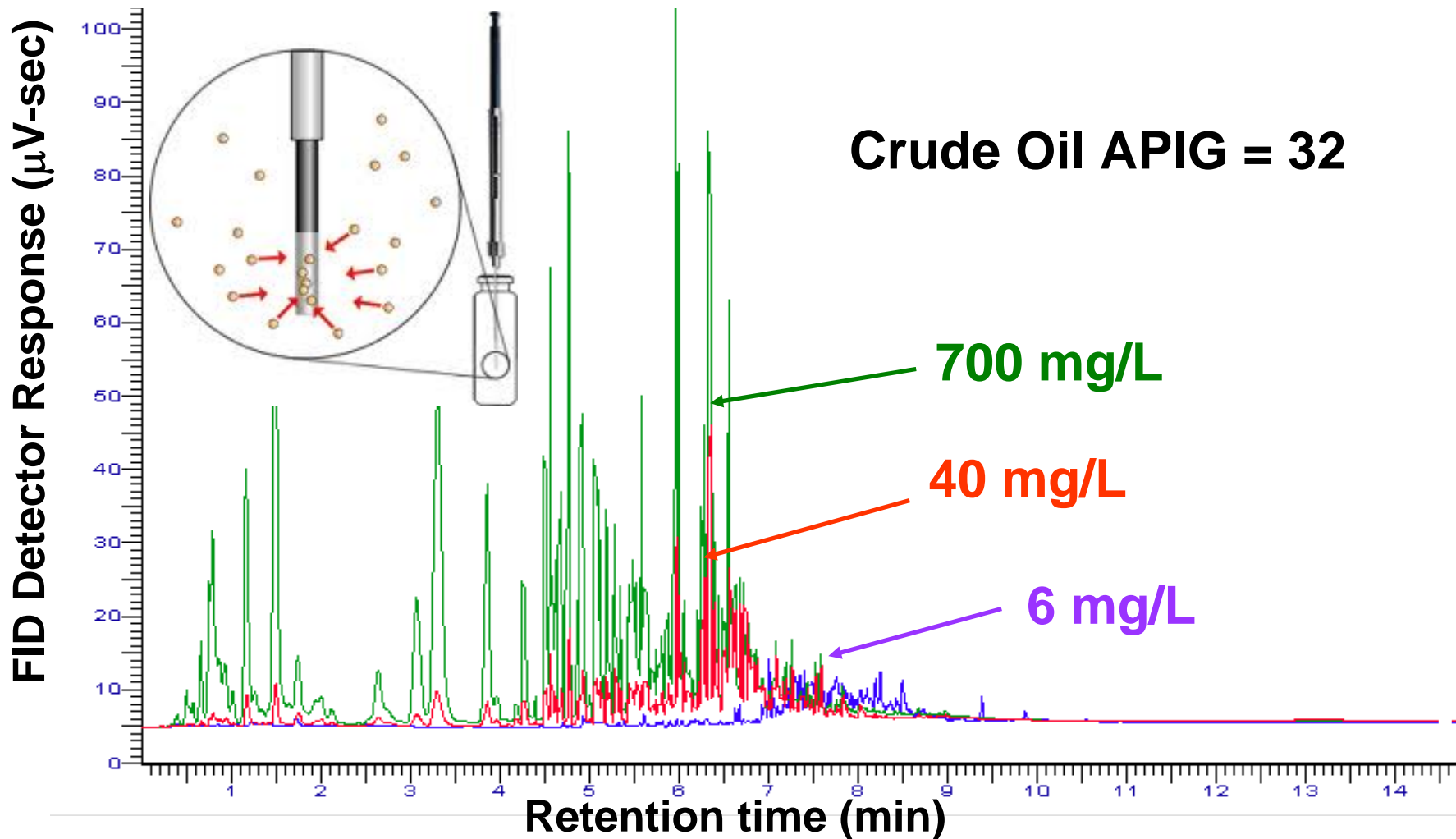
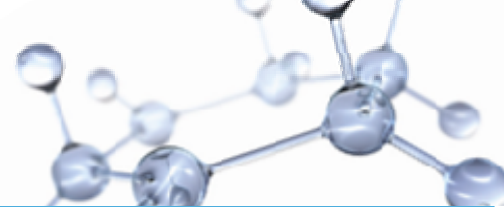
Single Hydrocarbon

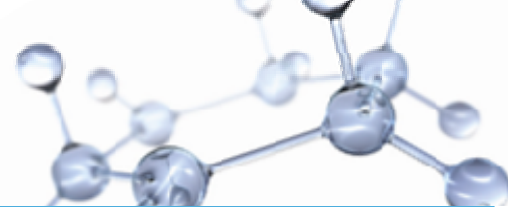


Multi-Component Oil



SPME Fiber Chromatograms for Crude Oil WAFs



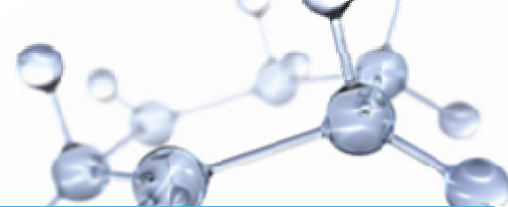


Outline of WAF Test (Cont'd)

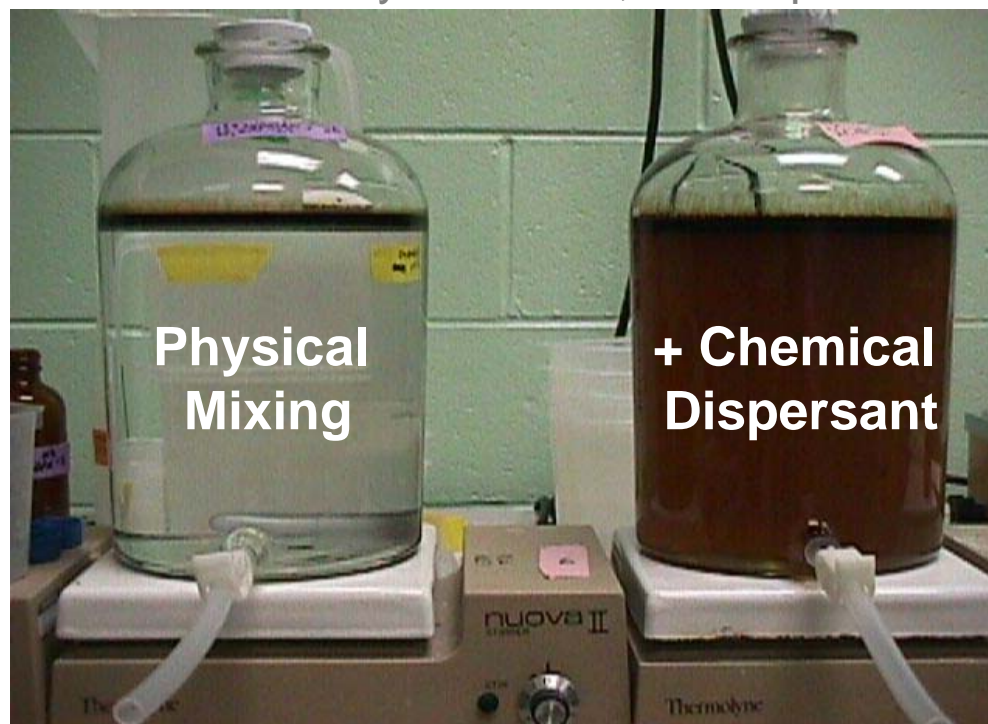


- Stop mixing / allow phase separation
 - Typically allow 1 hour unless adjustment to different temperature required (e.g. trout studies) which may require longer periods
- Withdraw solution from WAF test system
 - Discard first 100 mls
 - Collect sample for toxicity testing by directly transferring WAF via gravity flow to air tight exposure vessels to which test organisms are introduced
 - Need to consider oxygen depletion concerns especially for fish
 - + Use static renewal exposure design
 - + Add pure oxygen
 - Need to consider pH changes for algae
 - + Increase buffering capacity of test media
- Observe test organism response to WAFs

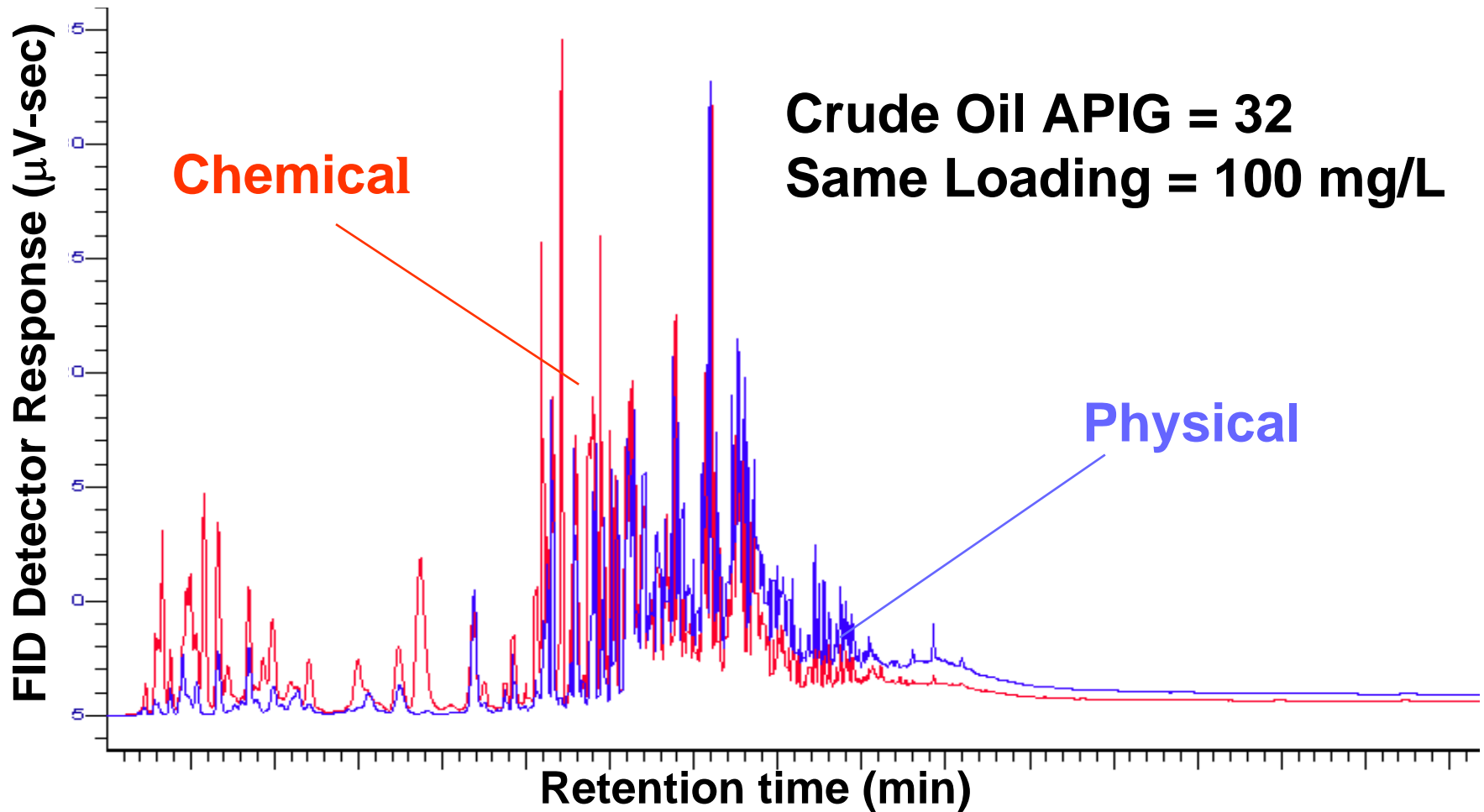
Chemical Dispersants



- Designed to exhibit low aquatic toxicity
 - Less toxic than the oil to be dispersed
- Increases amount of oil in aqueous test media
 - Augments “effective” loading potentially increasing dissolved or ‘bioavailable’ hydrocarbon concentrations
 - Increases undissolved hydrocarbon, i.e. droplets



SPME Chromatogram Comparison for Physical & Chemical Dispersion





Other Approaches

- Use of Water Soluble Fractions (WSF)
 - Filter WAF to remove undissolved oil
 - + Potential for removal of dissolved constituent
 - + Adds significant effort to test
 - + Can be used to investigate role of physical effects associated with highly dispersed WAFs
- Use of WAF / WSF dilutions
 - Prepare WAF / WSF at a given loading (e.g. 10 g oil /L water)
 - Make serial dilutions of the WAF / WSF
 - Exposure test organisms to WAF / WSF dilutions
 - Express toxicity in terms of % dilution
 - Traditionally used in oil spill studies

Cautionary Note: A 1:100 dilution of a 10g/L WAF \neq 100 mg/L WAF
since amount and composition of hydrocarbons will differ

Tools for Predicting Toxicity



- Additive Toxic Unit Model

- Given detailed composition of oil simulate composition of aqueous hydrocarbons in WAF test system
- Use TLM to calculate species-specific toxicity to all predicted hydrocarbons in WAF
- Calculate additive contribution of each hydrocarbon to toxicity

$$TU_i = \frac{C_{w,i}}{C_{w,i}^*} \quad \text{and} \quad \text{Total TU} = \sum_{i=1}^n TU_i$$

where:

$C_{w,i}$ = aqueous concentration of hydrocarbon i predicted in WAF

$C_{w,i}^*$ = aqueous effect concentration (e.g., LC_{50}) of hydrocarbon i

$TU < 0.3$

Toxicity Unlikely

$0.3 < TU < 2.0$

Toxicity Uncertain

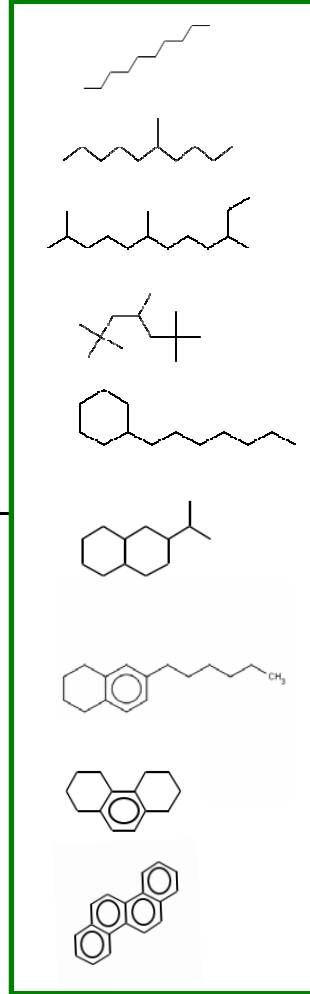
$TU > 2.0$

Toxicity Likely

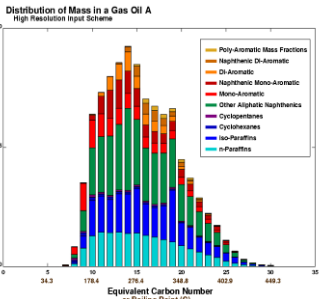


Overview of PETROTOX Model

Model Structure Library

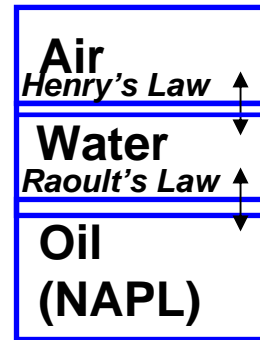


Petroleum Product Composition



Initial Petroleum Product Loading

WAF Model



WAF Hydrocarbon Concentrations

$C_{1, \text{water}}$
 $C_{2, \text{water}}$
 $C_{3, \text{water}}$

 $C_{n, \text{water}}$

$$\text{Toxic Units} = \sum_{i=1}^n (C_i / LC_{50i}) = 1?$$

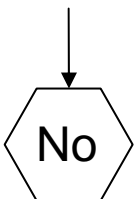
Effect Model



Hydrocarbon Toxicities

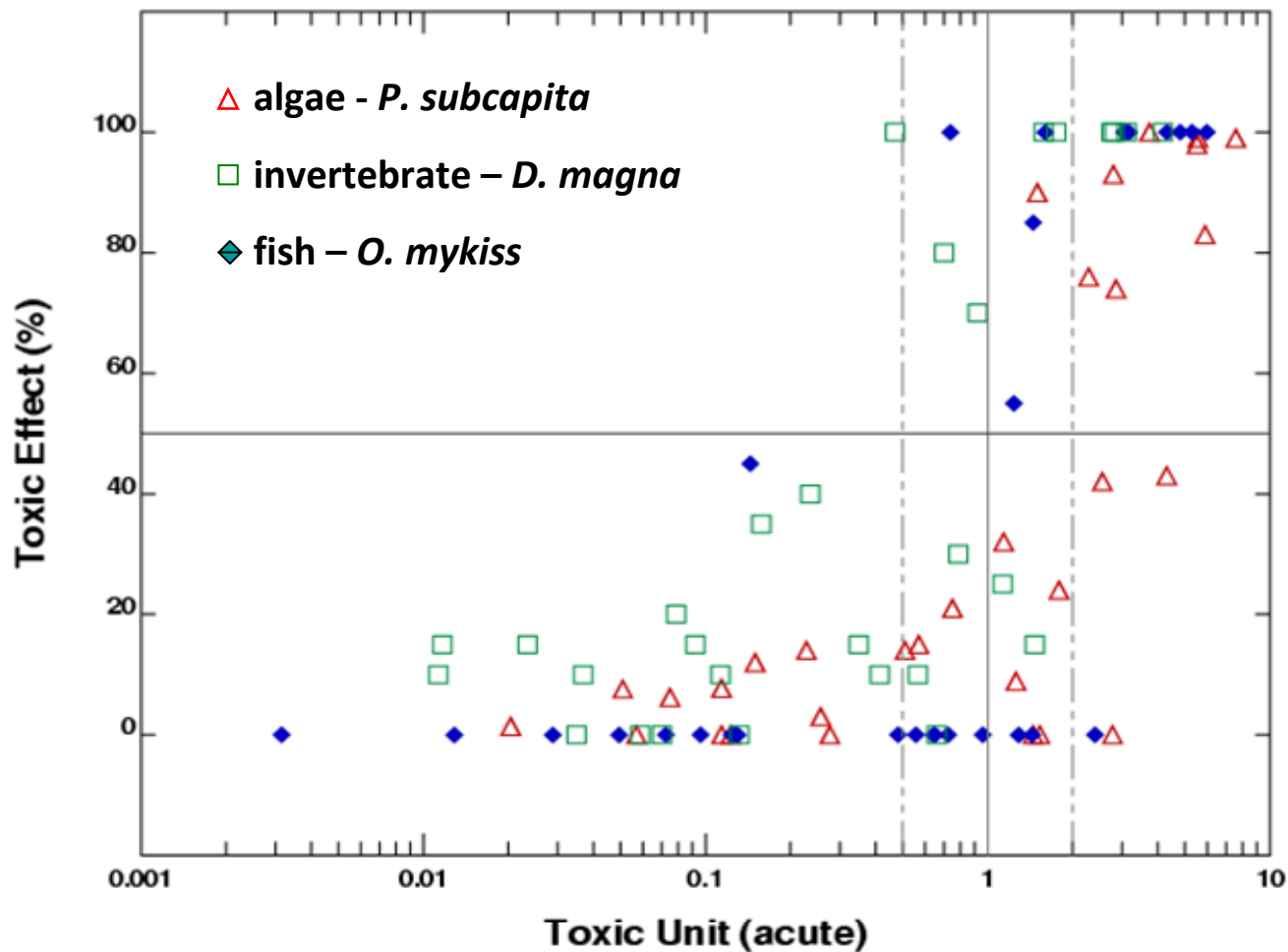
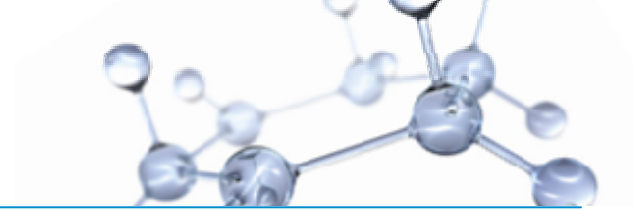
$LC_{50 1, \text{water}}$
 $LC_{50 2, \text{water}}$
 $LC_{50 3, \text{water}}$

 $LC_{50 n, \text{water}}$



Select new loading & repeat until convergence

Use of TLM to Predict Acute Toxicity of Gasolines



Source:
McGrath et al. (2005)
ET&C 24:2382–2394

Tools for Predicting Toxicity (Cont'd)



- Biomimetic Extraction Analysis:

- Ecotoxicity occurs when {molar} in organism lipid exceeds a critical threshold, i.e., CTLBB
- For given organism / endpoint, CTLBB is ~ constant for different hydrocarbons which act by a common mode of action
- Ecotoxicity of hydrocarbon mixtures is additive i.e., CTLBB concept applies to complex petroleum products
- SPME fibers serve as a surrogate for organism target lipid
- Total amount of hydrocarbons that sorb from a petroleum contaminated sample (e.g. WAF) to SPME fiber used for quantitative toxicity prediction

Mysid Toxicity Case Study



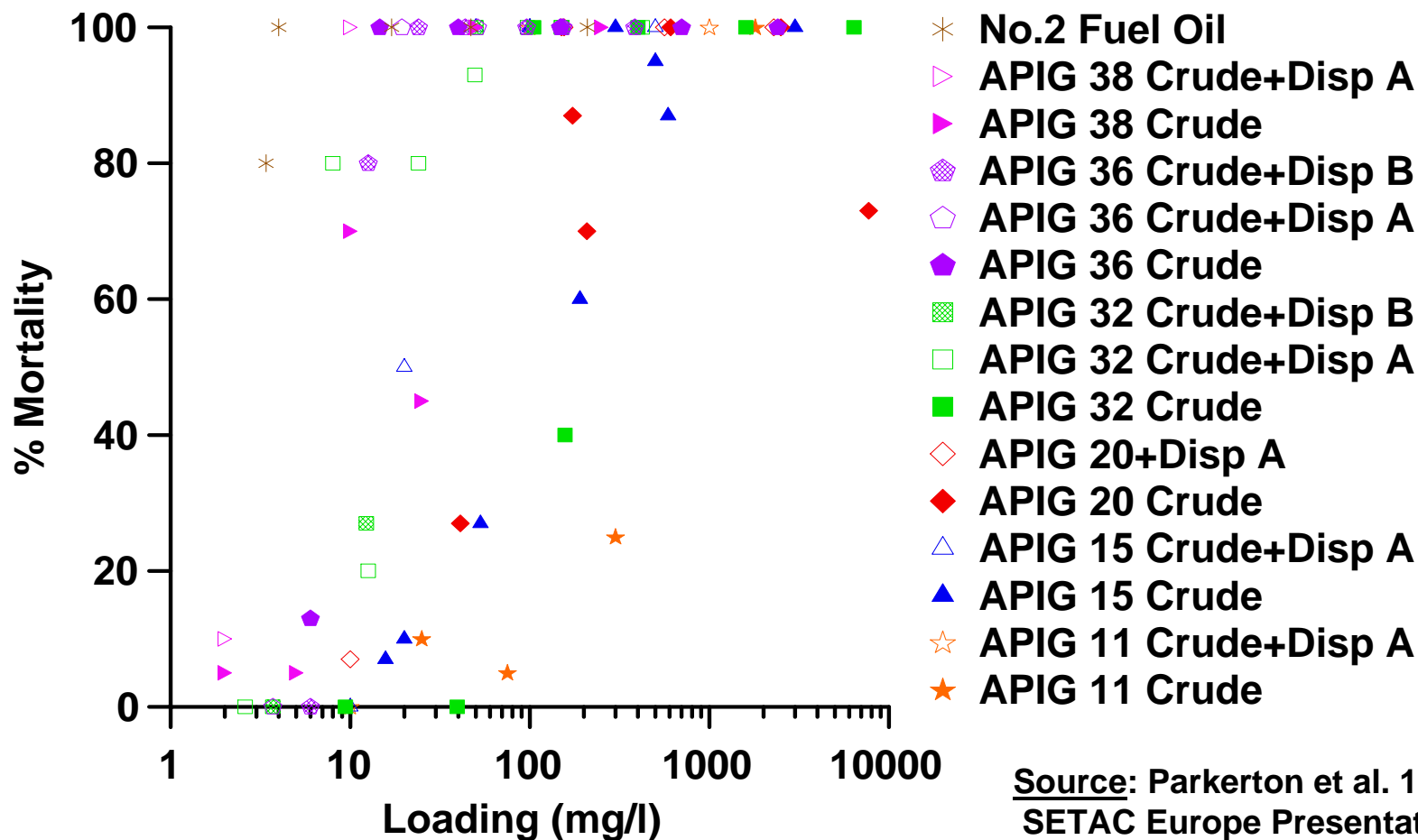
- Prepare physically and chemically dispersed WAFs
 - Five crude oils, no. 2 fuel oil
 - Two dispersants
 - Multiple oil loadings
- Measure SPME fiber concentrations associated with each WAF
 - Equilibrate fiber in WAF for 24 hrs
 - Inject fiber into GC/FID
 - Quantitate using molar response of C₂-naphthalene
 - Express results as $\mu\text{mol/ml PDMS} = \text{mM PDMS}$
- Determine 48-hr acute toxicity using *Mysidopsis bahia*





Mysid Toxicity vs Oil Loading

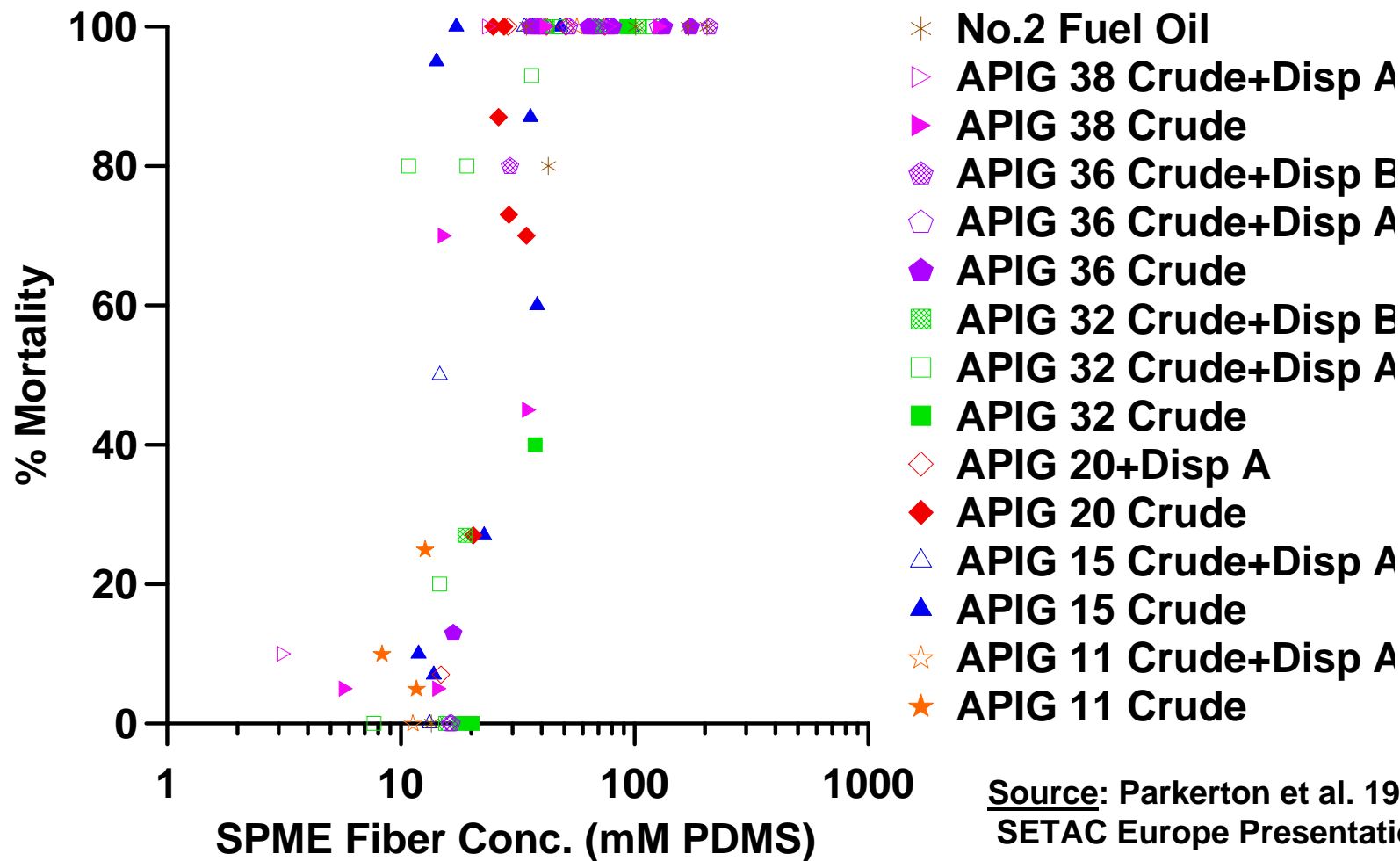
- Toxicity highly variable across treatments





Mysid Toxicity vs C_{Fiber}

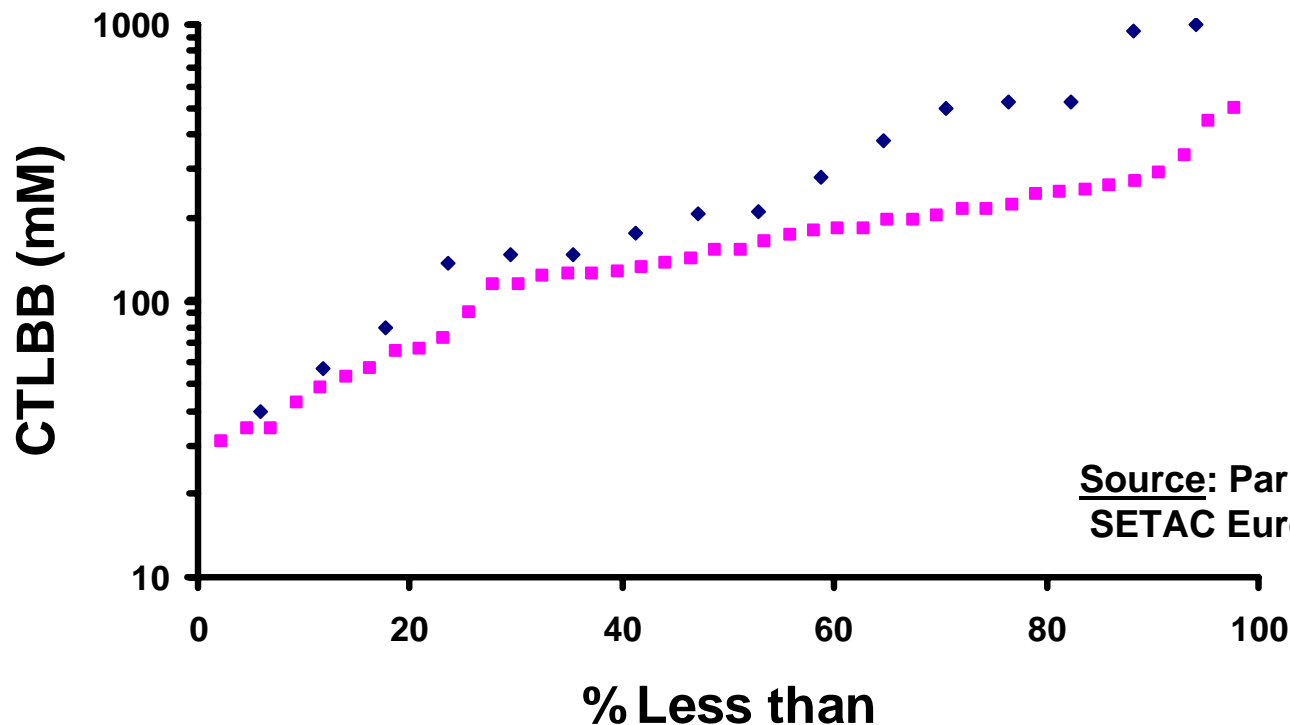
- Clear dose-response across treatments; dispersed oil not different





Further Validation Efforts

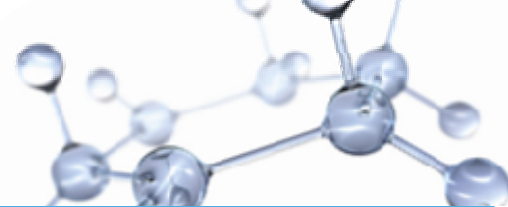
- Prepare WAFs using no. 2 fuel oil at different loadings
- Determine C_{Fiber} and toxicity for different test species
- Use C_{Fiber} – toxicity responses to estimate critical fiber burdens (CFBs)
- Translate CFBs into CTLBBs given $K_{\text{TL-W}} / K_{\text{PDMS-W}} \sim 8$



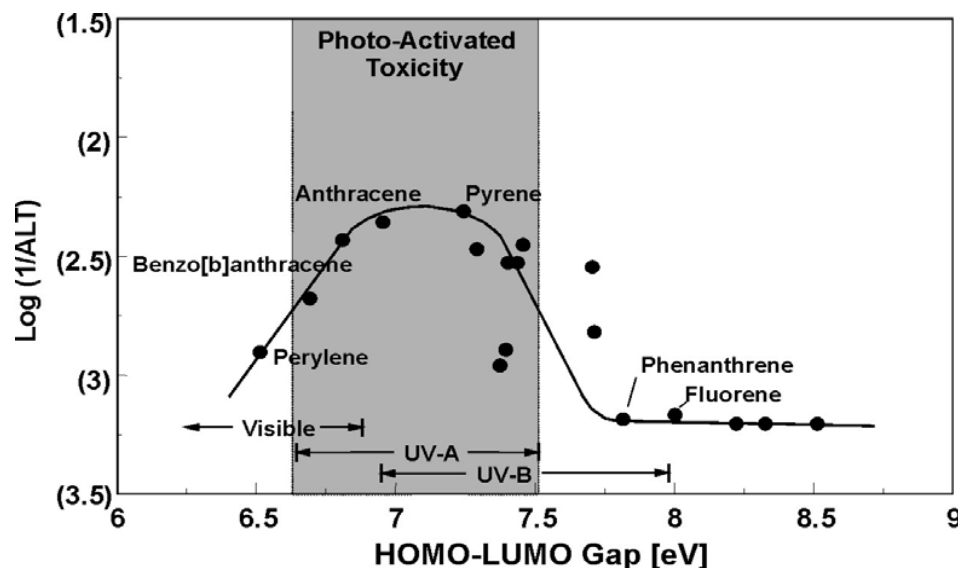
Source: Parkerton et al. 2009
SETAC Europe Presentation

◆ fuel oil SPME fibers ■ target lipid model

Photo-Enhanced Toxicity



- Selected PAHs shown to be more toxic in lab in presence of UV light



Mount et al., (2001)

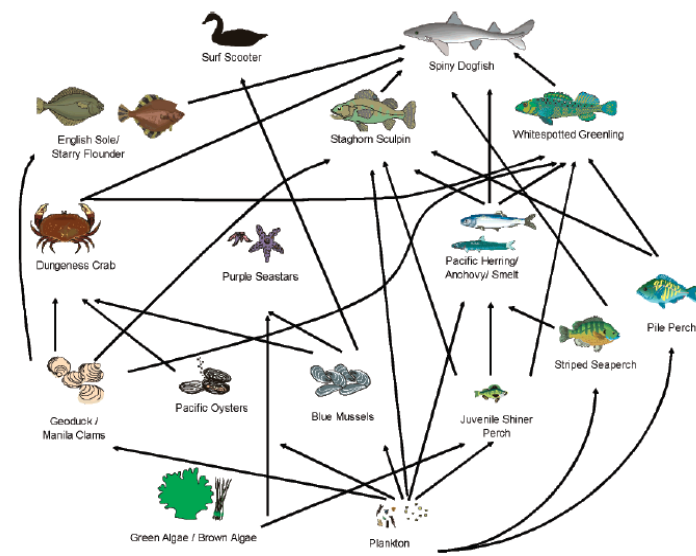
Linking exposure and dosimetry to risk from photo-activated toxicity of PAHs. Presented at the 2001 Annual SETAC Meeting. Baltimore, MD.

- Toxicity predicted by product of UV intensity and PAH tissue residue
 - UV intensity depends on location, season, time of day, water clarity; decreases exponentially with water depth
 - PAH tissue residue depends on PAH exposure concs and organism
- Influence of UV light on PAH toxicity offset by photodegradation
 - Estimated aqueous photolysis half-life for anthracene ca. minutes to days

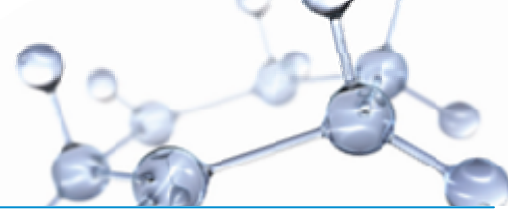
Bioaccumulation of PAHs in Foodchain



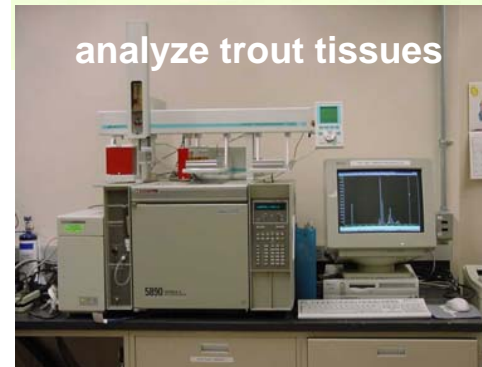
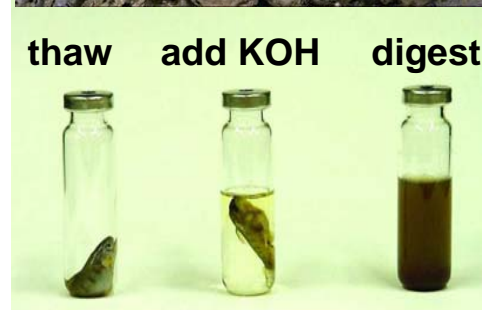
- Selected PAHs known to be carcinogenic/mutagenic, e.g. benzo(a)pyrene, dibenz(a,h,)anthracene, chrysene
- Bioconcentration at base of foodweb limited by dissolved PAH concs.
- Subsequent transfer to higher organisms mitigated by biotransformation processes
 - PAHs shown to biodilute, not biomagnify in foodweb
 - + Lab Biomagnification Factors (BMFs)
 - + Field Trophic Magnification Factors (TMFs)



Lab Dietary Bioaccumulation Test



- Spike hydrocarbons to commercial fish diet
 - Lipid content of diet 15%
 - Spike liquids directly, solids in corn oil
- Confirm dietary concentrations analytically
- Feed 3% ration of spiked diet to trout or carp (1-5 grams; 2-4% lipid) for 7 to 10 days (uptake)
- Transfer exposed fish to clean food (depuration)
- Analyze fish at different depuration times e.g. 0, 1, 3, 7, 14, 21 days
- Use hexachlorobenzene as positive control



Bioaccumulation Data Analysis

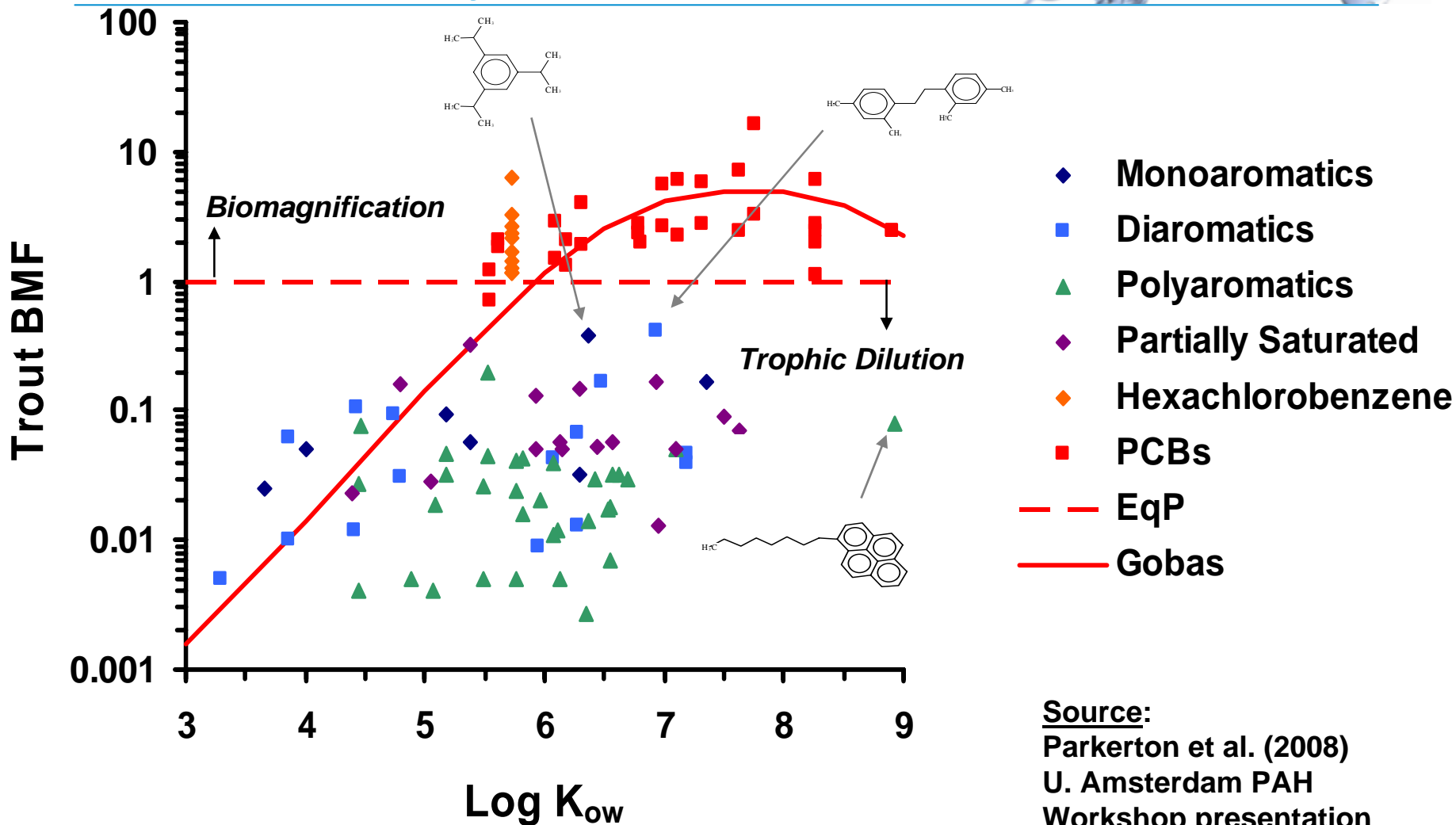


- Use experimental depuration data to deduce:
 - Growth-corrected half-life ($t_{1/2}$)
 - + Derived from slope of depuration plot & fish growth rate
 - Assimilation efficiency from diet (α)
 - + Derived from intercept of depuration plot & first-order model
 - Biomagnification factor (BMF)

$$BMF = \frac{C_{\text{fish, lipid}}}{C_{\text{diet, lipid}}} = \frac{\alpha I_{\text{diet}} t_{1/2}}{0.693} \frac{L_{\text{diet}}}{L_{\text{fish}}}$$

$BMF < 1$	Trophic Dilution
$BMF = 1$	Equilibrium Partitioning
$BMF > 1$	Biomagnification

Trout BMFs for Aromatic Hydrocarbons



PCB data from Fisk et al. 1988 ET&C 17:951



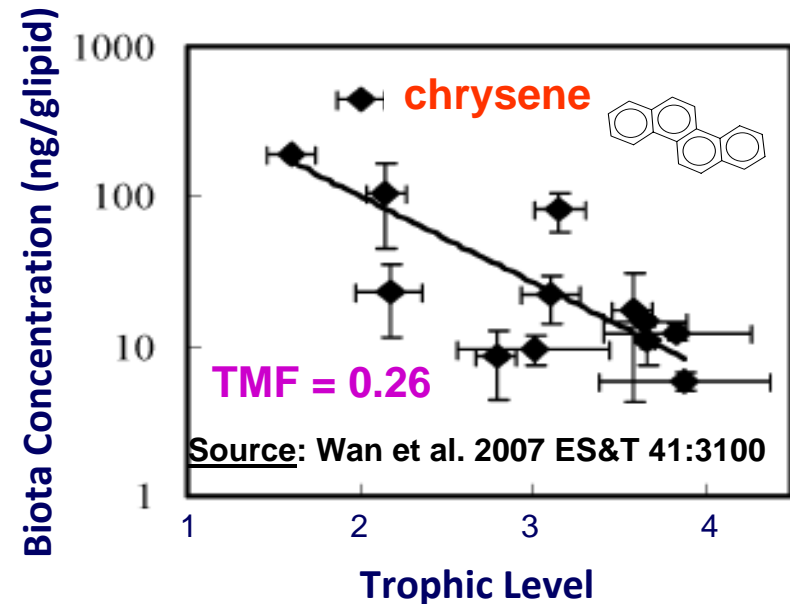
Field Bioaccumulation Assessment

- Collect field organisms from foodweb: analyse tissues for chemical and nitrogen isotopes
 - nitrogen isotopes used to determine trophic level (TL)
- Regress chemical concentration against TL to determine trophic magnification factor (TMF)
 - mean increase (biomagnification) or decrease (biodilution) of chemical / TL

$$\text{Log } C_{\text{lipid}} = a + b (\text{Trophic Level})$$

$$TMF = 10^b$$

$TMF < 1$	Trophic Dilution
$TMF = 1$	Equilibrium Partitioning
$TMF > 1$	Biomagnification





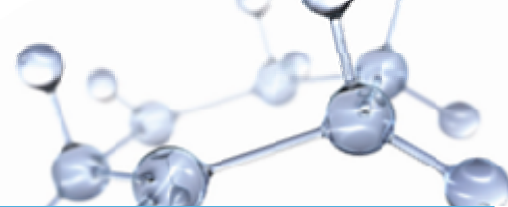
Literature TMFs for PAHs

PAH	TMF Ref =1	TMF Ref =2	TMF Ref =3
benz[a]anthracene	0.20	0.75	0.83
benzo[a]pyrene	0.24	0.75	0.80
benzo[e]pyrene	0.25	0.86	0.57
benzofluoranthene	0.27	0.84	0.69
benzo[ghi]perylene	0.66	0.75	0.72
chrysene	0.26	0.66	0.65
fluoranthene	0.11	0.72	0.60
indeno-123-cd]pyrene	0.81	0.75	0.80
dibenz[ah]anthracene	0.85		
perylene	0.24	0.67	0.77
phenanthrene	0.43	0.82	0.75
pyrene	0.17	0.74	0.62

Ref =1 Wan Y, Jin X, Hu J, Jin F. (2007). Trophic dilution of polycyclic aromatic hydrocarbons (PAHs) in a marine food web from Bohai Bay, North China. *Environ Sci Technol* 41:3109-3114.

Ref =2 Nfon, E., Cousins, I.T., et al. (2008). Biomagnification of organic pollutants in benthic and pelagic marine food chains from the Baltic Sea. *Sci Total Environ* 397:190-204.

Ref =3 Takeuchi, I., Miyoshi, N., et al. (2009). Biomagnification profiles of polycyclic aromatic hydrocarbons, alkylphenols and polychlorinated biphenyls in Tokyo Bay elucidated by d13C and d 15N isotope ratios as guides to trophic web structure. *Mar Poll Bull* 58:663-671.



Summary

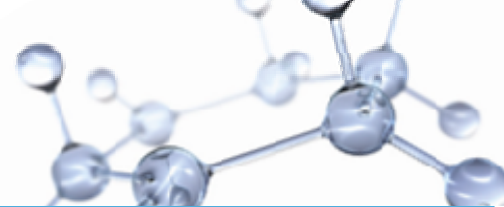
- The target lipid model provides a quantitative framework for predicting the acute and chronic toxicity of single and complex hydrocarbons
- The WAF test procedure is the preferred test method for assessing the aquatic toxicity of complex petroleum substances
 - method endorsed by OECD
 - accounts for multi-component dissolution behavior
- Passive sampling methods (e.g. SPME fibers) that quantify dissolved hydrocarbons in WAFs provide simple analytical tool to support testing and toxicity prediction
- Chemical dispersants exhibit low toxicity but can increase the bioavailability of hydrocarbons in the oil being dispersed
 - can result in increased WAF toxicity in lab studies
 - offset by role bioavailability plays in reducing field exposures, e.g. dilution, biodegradation
- Photo-enhanced toxicity and bioaccumulation in foodweb depends on dissolved PAH concentrations in the field; significance further limited by:
 - UV attenuation in water column and photodegradation
 - biodilution in the foodchain



Research Needs ?

- Develop reliable CTLBBs and ACRs for additional GOM species, e.g. sponges, corals for which limited data are available
- Develop data and improved models for characterizing toxicity of aromatic hydrocarbons on survival, growth and reproduction of key GOM species under time-variable exposure and field conditions, e.g. temperature, UV light, oxygen
- Link toxicity and population models to predict population-level responses
- Further investigate analytical and short-term toxicity screening tests for use in future spill response

Selected Publications



- Parkerton, T.F. A. D. Redman, J. A. McGrath, D. K. Letinski, E. J. Febbo, R. G. Manning, M. H. Comber, D. M. Di Toro (2011) Extension and validation of the target lipid model for deriving predicted no effect concentrations for hydrocarbons, Submitted to *Environ. Toxicol. Chem.*
- Burkhard, L.P., J.A. Arnot, M. R. Embry, K. J. Farley, R. A. Hoke, M. Kitano, H. A. Leslie, G. R. Lotufo, T. F. Parkerton, K. G. Sappington, G. T. Tomy (2011), Comparing laboratory and field measured bioaccumulation endpoints, *Integrated Environmental Assessment & Management*. DOI: 10.1002/ieam.244
- McElroy, A.E., M. G. Barron, N. Beckvar, S. B. Kane Driscoll, J. P. Meador, T. F. Parkerton, T. G. Preuss, J. A. Steevens (2011). A review of the tissue residue approach for organic and organometallic compounds in aquatic organisms, *Integrated Environmental Assessment & Management* 7:50-74.
- Hook, S.E., M. A. Lampi, E. J. Febbo, J. A. Ward, T. F. Parkerton (2010). Hepatic gene expression in rainbow trout (*Oncorhynchus mykiss*) exposed to different hydrocarbon mixtures, *Environ. Toxicol. Chem.* 29:2034-2043.
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- Arnot, J., D. Mackay, T. F. Parkerton, R. Zaleski, C. S. Warren (2010) Multimedia modeling of human exposure to chemical substances: the role of food web biomagnification and biotransformation, *Environ. Toxicol. Chem.* 29:44-55.
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- Arnot, J., D. Mackay, T.F. Parkerton, M. Bonnell (2008). A database of fish biotransformation rates, *Environ. Toxicol. Chem.* 27:263–2270.
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- Foster, K.I., D. Mackay, T.F. Parkerton, E. Webster, L. Milford (2005). A five stage risk assessment strategy for mixtures: gasoline as a case study, *Environ. Sci. Technol.* 39:2711-2718.
- McGrath JA, Hellweger FL, Parkerton TF, Di Toro DM. (2005). Application of the narcosis target lipid model to complex mixtures using gasolines as a case study, *Environ. Toxicol. Chem.* 24:2382-2394
- McGrath, J.A., T.F. Parkerton, and D.M. Di Toro (2004). Application of the narcosis target lipid model to algal toxicity and deriving predicted no effect concentrations. *Environ. Toxicol. Chem.* 23:2503–2517.